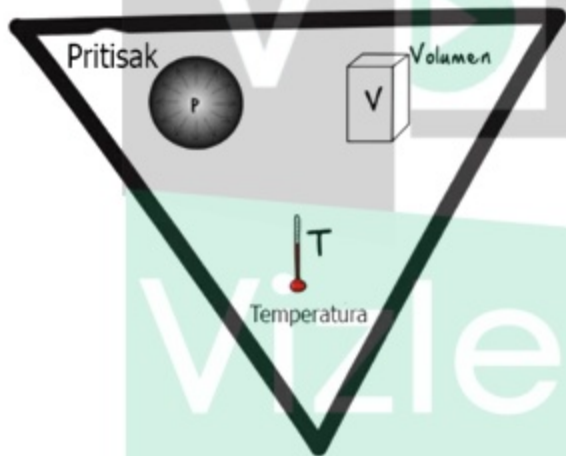


Gasni zakoni





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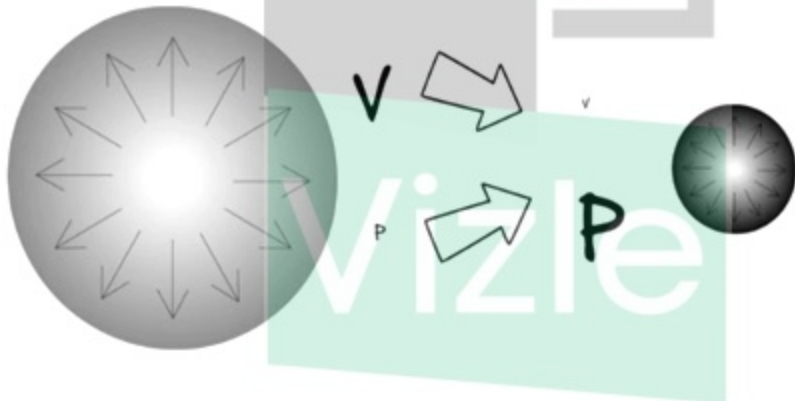
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Boyle i Mariotte

(T konstanto)

$$p \cdot V = \text{konstant}$$



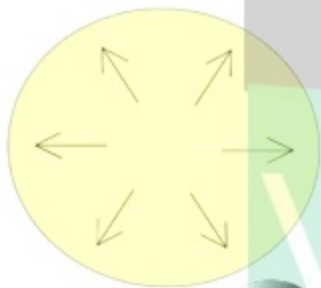


~~Celsius~~

Guillaume Amontons

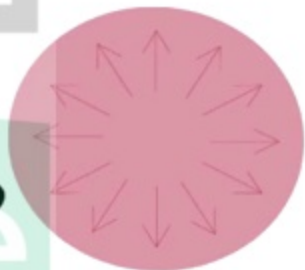
Kelvin-Skala

$$(V \text{ konstant})$$
$$\frac{T}{P} = \text{konstant}$$



T → T

P → P



Tegla

Gay-Lussac

(p konstant)

$$\frac{V}{T} = \text{konstant}$$



T ↑

V ↑

le

$$\frac{p \cdot V}{T} = \textit{konstant} \quad p \cdot V = \textit{konstante} \cdot T$$

$$p \cdot V = n \cdot R \cdot T$$

Količina supstance

Gasna konstanta

$$R \approx 8,314 \text{ J} \cdot \text{mol}^{-1} \cdot \text{K}^{-1}$$



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